

Identification of Atenolol and Chlorthalidone in Tablets by TLC acc. to USP and Quantification Using the TLC Explorer

Valanka D'Silva

R&D APAC Lab, Jigani, Bangalore, India

Abstract

This study presents a method for the identification of atenolol and chlorthalidone in tablet formulations using Thin Layer Chromatography (TLC) as per the USP monograph. The TLC Explorer system was employed for enhanced digital documentation and automated evaluation, facilitating also quantitative analysis and by that extended the use of TLC for this application. Results indicate that the test samples exhibited $R_{\rm f}$ values comparable to standard solutions, confirming identity. Quantitative assessments yielded concentrations within 97.6% to 102.6% of nominal values, indicating the method's applicability. This application underscores the TLC Explorer's efficacy in pharmaceutical TLC analysis and documentation.

Introduction

Atenolol and chlorthalidone combination tablets are used to treat high blood pressure (hypertension). Atenolol (2-[p-[2-hydroxy-3-(isopropylamino) propoxy]phenyl]acetamide) is a cardio-selective beta blocker that works by affecting the response to some nerve impulses in the heart. As a result, the heart beats slower and decreases the blood pressure. Chlorthalidone (2-chloro-5-(1-hydroxy-3-oxoisoindolin1-yl)benzene sulfonamide) is a diuretic that reduces the amount of water in the body by increasing the flow of urine, which helps to lower blood pressure.

Figure 1. Chemical structures of (A) atenolol and (B) chlorthalidone.

The United States Pharmacopeia (USP) monograph for atenolol and chlorthalidone tablets lists thin layer chromatography (TLC) as one of the methods for the identification test.² In many pharmacopeial methods, TLC is a frequently cited technique for identity testing. HPTLC, a high-performance version of TLC and often used with automation, is a robust, reliable, rapid, and inexpensive tool used in qualitative and quantitative analysis of pharmaceutical compounds. This technique delivers chromatographic separations/fingerprints that can be visualized for identification/quantification and saved as electronic images for documentation.³⁻⁴

In this application note, the USP monograph specified identification test of atenolol and chlorthalidone in tablets by TLC is performed on the new TLC Explorer documentation system (**Figure 2**) that is also used to extend this method to a quantitative evaluation by video densitometry.



Figure 2. TLC Explorer System



The TLC Explorer documentation system enables the digital and automated evaluation of TLC plates, enhancing the efficiency and accuracy of thin layer chromatography analyses. The device offers three illumination modes using LED light sources – white light (VIS), UV-A (366 nm), and UV-C (254 nm) – for the detection and fast measurement of the compounds of interest. The software offers special features like automated track recognition, simultaneous measurement of multiple plates and background signal correction. Overall, the TLC Explorer offers accurate TLC imaging for reliable densitometric measurements, enabling quantitative analysis and reliable data interpretation.

(read more at SigmaAldrich.com/tlc-explorer)

Experimental

Reagent Preparation

Mobile phase: Dissolve 6.80 mL of ammonia solution 25% in 100 mL water to obtain 1 N ammonium hydroxide. Mix 40 mL of this solution with 200 mL n-butyl alcohol to obtain a mixture of n-butyl alcohol and 1 N ammonium hydroxide (5:1, v:v).

Standard Preparation

Atenolol

- Standard stock solution 1 (50 mg/mL of atenolol):
 Weigh and dissolve 125 mg of atenolol in 2.5 mL of methanol.
- Standard solution 1 (20 mg/mL atenolol): Dilute 400 μL of standard stock solution 1 to 1000 μL with methanol.

Chlorthalidone

- Standard stock solution 2 (25 mg/mL chlorthalidone):
 Weigh and dissolve 62.5 mg of chlorthalidone in
 2.5 mL of methanol.
- Standard solution 2 (10 mg/mL chlorthalidone): Dilute 400 μ L of standard stock solution 2 to 1000 μ L with methanol.

Standard mix solutions I-VI: Prepare a total of six standard solutions (nos. I-VI) by pipetting 40 $\mu L, 100~\mu L, 200~\mu L, 300~\mu L, 400~\mu L$ and 440 μL each of standard stock solution 1 and standard stock solution 2 into six separate 2 mL vials. Add methanol to get to final volume of 1 mL. The resulting solutions contain 2.0, 5.0, 10.0, 15.0, 20.0, and 22.0 mg/mL of atenolol and 1.0, 2.5, 5.0, 7.5, 10.0, and 11.0 mg/mL of chlorthalidone, respectively.

Sample Preparation

As sample, tablets were used with a atenolol to chlorthalidone content ratio of 1:2.

Test solution I (for identification): Weigh and shake a quantity of powdered combination tablets, equivalent to 50 mg of atenolol (corresponding to 25 mg chlorthalidone), with 2.5 mL of methanol for 15 minutes and filter through 0.45 μm PVDF membrane. Nominal resulting concentration of 20 mg/mL atenolol and 10 mg/mL chlorthalidone.

Test solution II (for quantification): Weigh and shake a quantity of powdered combination tablets, equivalent to 25 mg of atenolol (corresponding to 12.5 mg chlorthalidone), with 2.5 mL of methanol for 15 minutes and filter through 0.45 μm PVDF membrane. Nominal resulting concentration of 10 mg/mL atenolol and 5 mg/mL chlorthalidone.

Recovery test solutions I-III: Prepare three samples by shaking a quantity of powdered tablets, equivalent to 25 mg of atenolol (corresponding to 12.5 mg chlorthalidone), spiked with 125 μ L, 250 μ L and 500 μ L each of standard stock solution 1 and standard stock solution 2 for 15 minutes. Add methanol to get to final volume of 2.5 mL and filter the solutions.

Instrument Parameters

Table 1. TLC conditions

| TLC parameters | | | |
|---------------------|---|--|--|
| Plate: | Silica Gel 60G F ₂₅₄ , 20 x 20 cm (1.00390) | | |
| Sample application: | 10 μL of each solution, manual application | | |
| Mobile phase: | n-Butyl alcohol:1 N ammonium hydroxide (5:1, v:v) | | |
| Chamber conditions: | Twin trough chamber with chamber saturation | | |
| Migration distance: | 17 cm | | |
| Drying: | Air-drying | | |
| Detection: | UV, 254 nm and 366 nm | | |

Results

The identification of atenolol and chlorthalidone in tablets performed according to USP monograph on the TLC Explorer under UV 254 nm and UV 366 nm is demonstrated in **Figure 3**. It additionally shows the calibration and recovery experiments performed on the TLC Explorer. **Table 2** summarizes the obtained chromatographic results ($R_{\rm f}$ values). As required by the USP monograph for the identification, the principal spots obtained from the test solution (track 3) correspond to the $R_{\rm f}$ value size and intensity of the respective standards solutions (tracks 1 & 2).

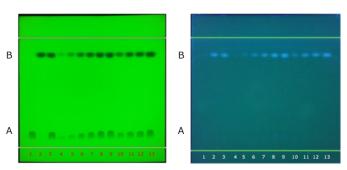


Figure 3. TLC chromatogram demonstrating the identification studies (track 1-3), calibration studies (tracks 4-9), test sample (10) and recovery studies (tracks 11-13) of atenolol (A) and chlorthalidone (B) in tablets under UV 254 nm (left) and UV 366 nm (right) by the TLC Explorer.

Table 2. Chromatographic data observed for standard solutions and test solutions under UV 254 nm by the TLC Explorer

| Track No. | Solution name | Concentration of Atenolol (µg/spot) | Concentration of Atenolol in applied solution (mg/mL) | R _f Atenolol | Concentration of Chlorthalidone (µg/spot) | Concentration of Chlorthalidone in applied solution (mg/mL) | <i>R_r</i> Chlorthalidone |
|--------------|----------------------------|---|--|----------------------------|---|--|--|
| 1 | Standard solution 1 | 200 | 20.0 | 0.122 | n.a.* | n.a.* | n.a.* |
| 2 | Standard solution 2 | n.a.* | n.a.* | n.a.* | 100 | 10.0 | 0.845 |
| 3 | Test solution I | 200# | 20.0 | 0.124 | 100# | 10.0 | 0.841 |
| 4 | Standard mix solution I | 20 | 2.0 | 0.092 | 10 | 1.0 | 0.848 |
| 5 | Standard mix solution II | 50 | 5.0 | 0.104 | 25 | 2.5 | 0.851 |
| 6 | Standard mix solution III | 100 | 10.0 | 0.114 | 50 | 5.0 | 0.847 |
| 7 | Standard mix solution IV | 150 | 15.0 | 0.121 | 75 | 7.5 | 0.847 |
| 8 | Standard mix solution V | 200 | 20.0 | 0.129 | 100 | 10.0 | 0.844 |
| 9 | Standard mix solution VI | 220 | 22.0 | 0.133 | 110 | 11.0 | 0.845 |
| 10 | Test solution II | 100# | 10.0 | 0.122 | 50# | 5.0 | 0.848 |
| 11 | Recovery test solution I | 125 | 12.5 | 0.124 | 62.5 | 6.25 | 0.849 |
| 12 | Recovery test solution II | 150 | 15.0 | 0.127 | 75 | 7.5 | 0.852 |
| 13 | Recovery test solution III | 200 | 20.0 | 0.139 | 100 | 10.0 | 0.848 |

[#] nominal values; *n.a.: not applicable

Calibration

The results of the calibration experiments utilizing standard mix solutions I-VI and the calibration function of the TLC Explorer unit are displayed in Figures 4 & 5 and Table 3 (for atenolol), and Figures 6 & 7 and Table 4 (for chlorthalidone).

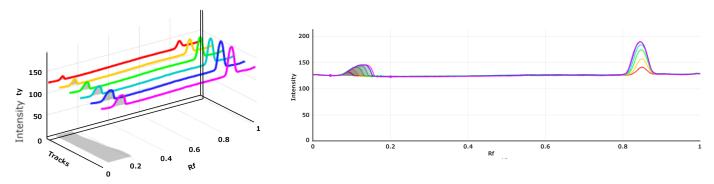


Figure 4. 3D densitogram (left, tracks 4 to 9) and 2D densitogram (right) demonstrating the calibration studies for atenolol utilizing standard mix solutions I-VI.

Table 3. Calibration data obtained for atenolol (20.0-220.0 µg/Spot), 6 calibrants

| Conc. (µg/spot) | Area Response |
|-----------------|---------------|
| 20 | 291.8 |
| 50 | 624.3 |
| 100 | 1064.9 |
| 150 | 1524.3 |
| 200 | 1997.7 |
| 220 | 2164.2 |
| Slope | 9.2898 |
| STEYX | 20.3671 |
| LOD (µg/mL) | 7.23 |
| LOQ (µg/mL) | 21.92 |

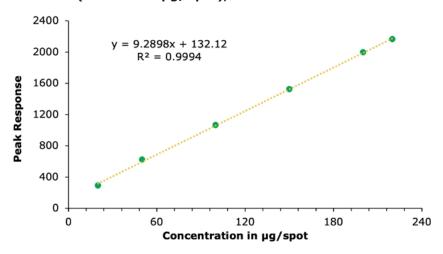
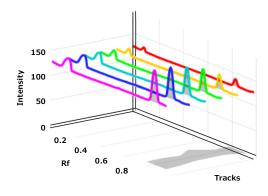


Figure 5. Calibration curve obtained for atenolol using standard mix solutions I-VI (c = 20.0, 50.0, 100.0, 150.0, 200.0 and 220.0 μ g/spot).



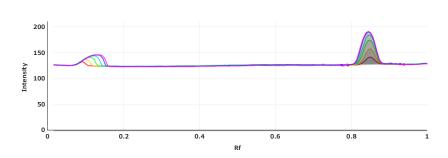


Figure 6. 3D densitogram (left, tracks 9 to 4) and 2D densitogram (right) demonstrating the calibration studies of chlorthalidone utilizing standard mix solutions I-VI.

Table 4. Calibration data obtained for chlorthalidone (20.0-220.0 µg/Spot), 6 calibrants

| Conc. (µg/spot) | Area Response |
|-----------------|---------------|
| 10 | 959.6 |
| 25 | 1631.6 |
| 50 | 2694.5 |
| 75 | 3683.5 |
| 100 | 4607.1 |
| 110 | 4932.4 |
| Slope | 39.7473 |
| STEYX | 71.8758 |
| LOD (µg/mL) | 5.97 |
| LOQ (µg/mL) | 18.08 |

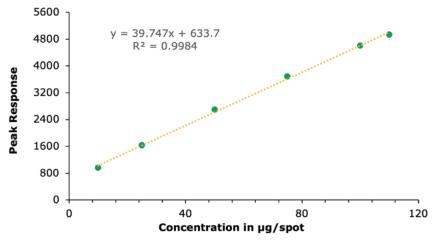


Figure 7. Calibration curve obtained for chlorthalidone using standard mix solutions I-VI (c = 10.0, 25.0, 50.0, 75.0, 100.0 and $110.0 \mu g/spot$)

The determined sensitivities LODs and LOQs were 7.23 and 21.92 $\mu g/spot$ for atenolol and 5.97 and 18.08 $\mu g/spot$ for chlorthalidone respectively. These represent concentrations in the applied sample solutions (10 μ L) of 0.72 and 2.19 mg/mL for atenolol and of 0.59 and 1.80 $\mu g/mL$ chlorthalidone.

Recovery

The results of the spike recovery experiments utilizing recovery test solution I-III are displayed in **Table 5** and **Table 6** and were found to be in the range of 90.03 to 98.6%.

Table 5. % Recovery determined for atenolol using recovery test solution I-III

| Sample | Spike concentration (µg/spot) | Determined spike concentration (µg/spot) | % Recovery |
|----------------------------|-------------------------------|--|------------|
| Recovery test solution I | 25 | 24.67 | 98.68 |
| Recovery test solution II | 50 | 46.04 | 92.08 |
| Recovery test solution III | 100 | 96.85 | 96.85 |

Table 6. % Recovery determined for chlorthalidone using recovery test solution I-III

| Sample | Spike concentration (µg/spot) | Determined spike concentration (µg/spot) | % Recovery |
|----------------------------|-------------------------------|--|------------|
| Recovery test solution I | 12.5 | 11.25 | 90.03 |
| Recovery test solution II | 25 | 24.21 | 96.83 |
| Recovery test solution III | 50 | 48.13 | 96.26 |

Results Test Sample

Quantification of the test sample II by applying the established calibration curve led to resulting concentrations of 97.64 μ g/spot for atenolol and 51.32 μ g/spot for chlorthalidone (**Table 7**), representing 97.64 % and 102.64 % of the respective nominal concentrations in the applied solutions (10 μ L) of 10 mg/mL atenolol and 5 mg/mL chlorthalidone.

Table 7. % Assay determined for atenolol and chlorthalidone in test solution II

| Sample | Area response | Determined concentration (µg/spot) | Nominal concentration (µg/spot) | % Assay |
|----------------|---------------|------------------------------------|---------------------------------|---------|
| Atenolol | 1039.2 | 97.64 | 100 | 97.64 |
| Chlorthalidone | 2673.5 | 51.32 | 50 | 102.64 |

Conclusion

A method was applied following the USP monograph specified identification test of atenolol and chlorthalidone in tablets using TLC. The assessment and the documentation of the chromatographic results were done with the TLC Explorer documentation system. The principal spots in the chromatogram obtained from the test solution were similar in R_f value, size, and intensity to the principal spot in the chromatogram obtained with the standard solution as required by the monograph. Additionally, beyond the USP monographs TLC use for identification, the method was extended to a quantitative assessment for the two analytes using the quantification option of the TLC Explorer. The quantification of the test sample resulted in 97.6 -102.6% agreement with the nominal concentrations.

This application demonstrates that the TLC Explorer documentation system is an efficient and practical tool for data collection and documentation, track identification, and $R_{\rm f}$ calculation as well as quantitation by video densitometry.

References

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| Description | Cat. No. |
|--|----------|
| Digital TLC Analysis and Documentation Device | |
| TLC Explorer, digital TLC analysis and documentation device | 1.52610 |
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| TLC Silica gel 60G F ₂₅₄ , glass plates 20 x 20 cm, Pk.25 | 1.00390 |
| Solvents, Reagents, Consumables & Reference Materials | |
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| Water, suitable for HPLC | 270733 |
| 1-Butanol, suitable for HPLC, ≥99.7% | 34867 |
| Ammonia solution 25%, for HPLC LiChropur™ | 5.43830 |
| Millex® HV Durapore (PVDF) 0.45 μm | SLHV033N |
| Atenolol, United States Pharmacopeia (USP) Reference Standard | 1044403 |
| Chlorthalidone, United States Pharmacopeia (USP) Reference Standard | 1130006 |

Related Products

| Description | Cat. No. |
|---|----------|
| Pharmacopeia & Metrological Institute Standards | |
| Atenolol, Pharmaceutical Secondary Standard; Certified Reference Material | PHR1909 |
| Atenolol, European Pharmacopoeia (EP) Reference Standard | A1340000 |
| Atenolol, British Pharmacopoeia (BP) Reference Standard | BP492 |
| Chlorthalidone, Pharmaceutical Secondary Standard; Certified reference material | PHR3404 |
| Chlorthalidone, European Pharmacopoeia (EP) Reference Standard | C1950000 |
| Chlorthalidone, British Pharmacopoeia (BP) Reference Standard | BP491 |

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