

**Enzymatic Assay of CATECHOL OXIDASE  
(EC 1.14.18.1)**

**PRINCIPLE:**

Catechol + O<sub>2</sub>  $\xrightarrow{\text{Catechol Oxidase}}$  O-Benzoquinone + H<sub>2</sub>O

O-Benzoquinone + Ascorbic Acid  $\longrightarrow$  Catechol + Dehydro-Ascorbic Acid

**CONDITIONS:** T = 25°C, pH = 6.5, A<sub>265nm</sub>, Light path = 1 cm

**METHOD:** Continuous Spectrophotometric Rate Determination

**REAGENTS:**

- A. 50 mM Potassium Phosphate Buffer, pH 6.5 at 25°C  
(Prepare 100 ml in deionized water using Potassium Phosphate, Monobasic, Prod. No. P-5379. Adjust to pH 6.5 at 25°C with 1 M NaOH.)
- B. 5 mM Catechol Solution  
(Prepare 10 ml in Reagent A using Catechol, Prod. No. C-9510. **PREPARE FRESH.**)
- C. 2.1 mM L-Ascorbic Acid Solution  
(Prepare 10 ml in Reagent A using L-Ascorbic Acid, Sodium Salt, Prod. No. A-7631. **PREPARE FRESH.**)
- D. 0.065 mM Ethylenediaminetetraacetic Acid Solution (EDTA)  
(Prepare 10 ml in Reagent A using Ethylenediaminetetraacetic Acid, Disodium Salt, Stock No. ED2SS.)
- E. Catechol Oxidase Enzyme Solution  
(Immediately before use, prepare a solution containing 500 - 1000 units/ml of Catechol Oxidase in cold Reagent A.)

**Enzymatic Assay of CATECHOL OXIDASE  
(EC 1.14.18.1)**

**PROCEDURE:**

Pipette (in milliliters) the following reagents into suitable quartz cuvettes:

	<u>Test</u>	<u>Blank</u>
Reagent A (Buffer)	2.60	2.80
Reagent B (Catechol)	0.10	0.10
Reagent C (Ascorbic Acid)	0.10	----- <sup>1</sup>
Reagent D (EDTA)	0.10	0.10

Mix by inversion and equilibrate to 25°C. Monitor the  $A_{265nm}$  until constant, using a suitably thermostatted spectrophotometer. Then add:

Reagent E (Enzyme Solution)	0.10	-----
-----------------------------	------	-------

Immediately mix by inversion and record the decrease in  $A_{265nm}$  for approximately 5 minutes. Obtain the  $r A_{265nm}/\text{minute}$  using the maximum linear rate for both the Test and Blank.

**CALCULATIONS:**

$$\text{Units/mg enzyme} = \frac{(r A_{265nm}/\text{min Test} - r A_{265nm}/\text{min Blank})}{(0.001) (\text{mg enzyme/RM})}$$

0.001 = The change in  $A_{265nm}/\text{minute}$  per unit of Catechol Oxidase at pH 6.5 at 25°C in a 3 ml reaction mix  
 RM = Reaction Mix (final volume = 3 ml)

**UNIT DEFINITION:**

One unit will cause the change in  $A_{265nm}$  of 0.001 per minute at pH 6.5 and 25°C in a 3 ml reaction mix containing catechol and L-ascorbic acid.

**FINAL ASSAY CONCENTRATION:**

In a 3 ml reaction mix, the final concentrations are 50 mM potassium phosphate, 0.17 mM catechol, 0.07 mM L-ascorbic acid and 50 - 100 units of catechol oxidase.

**Enzymatic Assay of CATECHOL OXIDASE  
(EC 1.14.18.1)**

**REFERENCES:**

Dawson, C.R., and Magee, R.J. (1955) *Methods in Enzymology* **II**, 817-821.

Marumo, K., and Waite, J.H. (1986) *Biochim. Biophys. Acta* **872**, 98-103.

**NOTES:**

1. Ascorbic acid is not included in the blank to produce a true blank. Possible interaction with catechol and ascorbic acid might give false readings.
2. All product and stock numbers, unless otherwise indicated, are Sigma product and stock numbers.

**This procedure is for informational purposes. For a current copy of Sigma's quality control procedure contact our Technical Service Department.**