7.75201.0004-xxxxxxxxx msp. September 2019

Supelco_®

1.14438.0001

MQuant® Iron Test



1. Method

Determination with color-card comparator

All iron ions are reduced to iron(II) ions. In a thioglycolate-buffered medium these react with a triazine derivative to form a red-violet complex. The iron concentration is measured **semiquantitatively** by visual comparison of the color of the measurement solution with the color fields of a color card.

2. Measuring range and number of determinations

Measuring range / color-scale graduation	Number of determinations	
0.2 - 0.4 - 0.6 - 0.8 - 1.0 - 1.3 - 1.6 - 2.0 - 2.5 mg/l Fe	500	

3. Applications

This test measures bivalent and trivalent iron in its dissolved form as well as fresh colloidal iron(III) hydroxide.

Sample material:

Groundwater and surface water, seawater Drinking water and mineral water Waters from aquaculture Boiler and boiler feed water, cooling water Industrial water Wastewater and percolating water Food after appropriate sample pretreatment

4. Influence of foreign substances

This was checked individually in solutions containing 1 and 0 mg/l Fe. The determination is not yet interfered with up to the concentrations of foreign substances given in the table. Cumulative effects were not checked; such effects can, however, not be excluded.

Con	Concentrations of foreign substances in mg/l or %						
Ca ²⁺ 10 Cd ²⁺ CN- Co ²⁺	000 50	Mg ²⁺ Mn ²⁺ NH ₄ +	0.5 1000 100 1000 1000 1000 1	S ²⁻ SCN- SiO ₃ ²⁻	500 1000 1000 1000	NaNO ₃	0 % 1 % 2 % 20 % 20 % 20 %

¹⁾ tested with nonionic, cationic, and anionic surfactants

5. Reagents and auxiliaries

Please note the warnings on the packaging materials!

The test reagent is stable up to the date stated on the pack when stored closed at +15 to +25 °C.

Package contents:

- 2 bottles of reagent Fe-1 (in aluminium container)
- 1 graduated 5-ml plastic syringe
- 2 test tubes with screw caps (in comparator block)
- 1 color card

Other reagents and accessories:
Nitric acid 65 % for analysis EMSURE®, Cat. No. 100456
MQuant® Universal indicator strips pH 0 -14, Cat. No. 109535
Sodium hydroxide solution 1 mol/l Titripur®, Cat. No. 109137
Hydrochloric acid 1 mol/l Titripur®, Cat. No. 109057 Iron standard solution Certipur®,1000 mg/l Fe, Cat. No. 119781

Refill pack:

Cat. No. 118458

Iron Test

Refill pack for 114759, 114438, and 114403

(Reagent without technical accessories for the number of determinations stated in section 2)

6. Preparation

- Analyze immediately after sampling. Otherwise preserve with nitric acid 65 % (1 ml nitric acid per 1 l of sample solution).
- The pH must be within the range 1 10. Adjust, if necessary, with sodium hydroxide solution or hydrochloric acid.
- · Filter turbid samples.

7. Procedure

Open the box and set up with both test tubes on the left.

Slide the comparator block all the way to the left, so that the end holding the test tubes protrudes laterally over the bottom part of the box

Unfold the color card and insert it, colored end first, into the slit at the lower right-hand edge of the box.

	Measurement sample tube nearer to the tester (A)	tube <u>farther from</u> the tester (B)					
Pretreated sample (10 - 40 °C)	5 ml	5 ml	Inject into the test tube with the syringe.				
Reagent Fe-1	3 drops ¹⁾	-	Add, close the tube, and mix.				

Leave to stand for 3 min (reaction time).

Slide the color card through to the left until the closest possible color match is achieved between the two open test tubes when viewed from above.

Read off the result in mg/l $\stackrel{\cdot}{\text{Pe}}$ from the color card at the lower right-hand edge of the comparator block within the bottom part of the box.

1) Hold the bottle vertically while adding the reagent!

Notes on the measurement:

- The color of the measurement solution remains stable for at least 60 min after the end of the reaction time stated above.
- Turbidity in the measurement solution makes the color comparison more difficult.
- If the color of the measurement solution is equal to or more intense than the darkest color on the scale, repeat the measurement using **fresh**, diluted samples until a value of less than 2.5 mg/l Fe is obtained.

Concerning the result of the analysis, the dilution must be taken into account:

Result of analysis = measurement value x dilution factor

8. Method control

To check test reagent, measurement device, and handling: Dilute the iron standard solution with distilled water to 1.0 mg/l Fe and analyze as described in section 7. Additional notes see under www.qa-test-kits.com.

9. Notes

- Reclose the reagent bottle immediately after use.
- Rinse the test tubes and the syringe with distilled water only.
- The test reagent must not be run off with the wastewater! Information on disposal can be obtained at www.disposal-testkits.com.

