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## **ProductInformation**

# Sodium bicarbonate Cell Culture Tested

Product Number **S 4019**Store at Room Temperature

#### **Product Description**

Molecular Formula: NaHCO<sub>3</sub> Molecular Weight: 84.01 CAS Number: 144-55-8

pK<sub>a</sub>: 6.37, 10.25 (carbonic acid)

Synonyms: sodium hydrogen carbonate, sodium acid

carbonate, baking soda

This Hybri-Max<sup>®</sup> product has been tested for suitability in hybridoma and cell culture research (2.2 mg/ml).

Sodium bicarbonate occurs in nature as the mineral nacholite. It is used in the production of many sodium salts and as a source of carbon dioxide.

Manufacturing applications of sodium bicarbonate include fire extinguishers and cleaning compounds. Bicarbonate is an important agent of pH regulation *in vivo* and the intracellular transport of sodium bicarbonate has been widely investigated. <sup>2,3,4</sup>

Sodium bicarbonate is used as a buffer in biochemistry, molecular biology, and chromatography. The use of sodium bicarbonate in cell culture media has been discussed. A review of methods of analysis for ochratoxin A from plasma and foods that includes the use of sodium bicarbonate has been published. A protocol has been reported on the separation of inorganic and organic arsenic species by HPLC in tandem with ICP mass spectrometry. The coppermediated cleavage of histidine-containing peptides in sodium bicarbonate buffer has been studied.

### **Precautions and Disclaimer**

For Laboratory Use Only. Not for drug, household or other uses.

### **Preparation Instructions**

This product is soluble in water (100 mg/ml), yielding a clear, colorless solution. At 25 °C, the pH of a freshly prepared 0.1 M aqueous solution of sodium bicarbonate is 8.3.

#### References

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- 8. Allen, G., and Campbell, R. O., Specific cleavage of histidine-containing peptides by copper(II). Int. J. Pept. Protein Res., **48(3)**, 265-273 (1996).

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