

1.14403.0001

MQuant® Iron Test

Fe

1. Method

Determination with color-card comparator

All iron ions are reduced to iron(II) ions. In a thioglycolate-buffered medium these react with a triazine derivative to form a red-violet complex. The iron concentration is measured **semiquantitatively** by visual comparison of the color of the measurement solution with the color fields of a color card.

2. Measuring range and number of determinations

Measuring range / color-scale graduation	Number of determinations
0.01 - 0.02 - 0.03 - 0.04 - 0.06 - 0.08 - 0.10 - 0.15 - 0.20 mg/l Fe	300

3. Applications

This test measures bivalent and trivalent iron in its dissolved form as well as fresh colloidal iron(III) hydroxide.

Sample material:

Groundwater and surface water, seawater
Drinking water and mineral water
Waters from aquaculture
Boiler and boiler feed water, cooling water
Industrial water
Wastewater and percolating water
Food after appropriate sample pretreatment

4. Influence of foreign substances

This was checked individually in solutions containing 0.08 and 0 mg/l Fe. The determination is not yet interfered with up to the concentrations of foreign substances given in the table. Cumulative effects were not checked; such effects can, however, not be excluded.

Concentrations of foreign substances in mg/l or %							
Al ³⁺	1000	Cu ²⁺	0.5	Pb ²⁺	10	EDTA	0 %
Ca ²⁺	1000	F ⁻	1000	PO ₄ ³⁻	500	Surfactants ¹⁾	1 %
Cd ²⁺	50	Hg ²⁺	100	S ²⁻	1000	Na-acetate	2 %
CN ⁻	10	Mg ²⁺	1000	SCN ⁻	1000	NaCl	20 %
Co ²⁺	1	Mn ²⁺	1000	SiO ₃ ²⁻	1000	NaNO ₃	20 %
CO ₃ ²⁻	1000	NH ₄ ⁺	1000	SO ₃ ²⁻	500	Na ₂ SO ₄	20 %
Cr ³⁺	10	Ni ²⁺	1	Zn ²⁺	500		
Cr ₂ O ₇ ²⁻	50	NO ₂ ⁻	10				

¹⁾ tested with nonionic, cationic, and anionic surfactants

5. Reagents and auxiliaries

Please note the warnings on the packaging materials!

The test reagent is stable up to the date stated on the pack when stored closed at +15 to +25 °C.

Package contents:

2 bottles of reagent Fe-1 (in aluminium container)
2 test tubes with screw caps (in comparator block)
1 color card

Other reagents and accessories:

Nitric acid 65 % for analysis EMSURE®, Cat. No. 1.00456
MQuant® Universal indicator strips pH 0 -14, Cat. No. 1.09535
Sodium hydroxide solution 1 mol/l Titripur®, Cat. No. 1.09137
Hydrochloric acid 1 mol/l Titripur®, Cat. No. 1.09057
Iron standard solution Certipur®, 1000 mg/l Fe, Cat. No. 1.19781

MQuant® Flat-bottomed long tubes with screw caps for MQuant® tests with color-card comparator (12 pcs), Cat. No. 1.14901

Refill pack:

Cat. No. 1.18458

Iron Test

Refill pack for 1.14759, 1.14438, and 1.14403

(Reagent **without technical accessories** for the number of determinations stated in section 2)

6. Preparation

- Analyze immediately after sampling. Otherwise preserve with nitric acid 65 % (1 ml nitric acid per 1 l of sample solution).
- The pH must be within the range 1 - 10.** Adjust, if necessary, with sodium hydroxide solution or hydrochloric acid.
- Filter turbid samples.

7. Procedure

Open the box and set up with both test tubes **on the left**.

Unfold the color card and insert it, colored end first, into the slit at the lower **right-hand** edge of the box.

	Measurement sample tube nearer to the tester (A)	Blank tube farther from the tester (B)	
Pretreated sample (10 - 40 °C)	20 ml	20 ml	Fill the test tube to the mark (= 20 ml).
Reagent Fe-1	5 drops ¹⁾	-	Add, close the tube, and mix.

Leave to stand for 3 min (reaction time).

Slide the color card through to the left until the closest possible color match is achieved between the two open test tubes when viewed from above.

Read off the result in mg/l Fe from the color card at the lower right-hand edge of the box.

¹⁾ Hold the bottle vertically while adding the reagent!

Notes on the measurement:

- The color of the measurement solution remains stable for at least 60 min after the end of the reaction time stated above.
- Turbidity in the measurement solution makes the color comparison more difficult.
- If the color of the measurement solution is equal to or more intense than the darkest color on the scale, repeat the measurement using **fresh**, diluted samples until a value of less than 0.20 mg/l Fe is obtained.

Concerning the result of the analysis, the dilution must be taken into account:

$$\text{Result of analysis} = \text{measurement value} \times \text{dilution factor}$$

8. Method control

To check test reagent, measurement device, and handling:

Dilute the iron standard solution with distilled water to 0.10 mg/l Fe and analyze as described in section 7.

Additional notes see under www.qa-test-kits.com.

9. Notes

- Reclose the reagent bottle immediately after use.
- Rinse the test tubes **with distilled water only**.
- The test reagent must not be run off with the wastewater!** Information on disposal can be obtained at www.disposal-test-kits.com.

