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ProductInformation

Cyclohexanone

Product Code **C10,218-0** Store at Room Temperature **Replacement for Product Number C 8390**

Product Description

Molecular Formula: $C_6H_{10}O$ Molecular Weight: 98.14 CAS Number: 108-94-1 Density: 0.9478 g/ml (25 °C)¹ Boiling point: 155 °C (760 torr)¹ Melting point: -32.1 °C¹ Synonyms: ketohexamethylene, pimelic ketone

Cyclohexanone is a solvent that is used in organic synthesis. It is obtained from cyclohexanol by catalytic dehydrogenation or by oxidation; the latter process gives adipic acid as an additional product. It may also be produced from cyclohexane by oxidation, which gives both cyclohexanone and cyclohexanol as products.¹

Notable uses of cyclohexanone include the production of adipic acid for nylon and of caprolactam. Cyclohexanone is also a solvent for cellulose acetate, nitrocelluose, natural resins, vinyl resins, polyvinyl chloride and its copolymers, methacrylate ester polymers, waxes, and fats.¹

Cyclohexanone has been investigated in the potential use of ionic liquids in liquid membranes for the selective transport of organic molecules.² Studies of the nanocatalyst-mediated conversion of cyclohexanone to its oxime and caprolactam have been published.^{3,4}

Precautions and Disclaimer

For Laboratory Use Only. Not for drug, household or other uses.

Preparation Instructions

This product is soluble in ethanol (0.1 ml/ml, 10% v/v), yielding a clear, colorless solution. Cyclohexanone is generally miscible with ether and other common organic solvents. This product is also soluble in water (87 mg/ml).¹

References

- 1. The Merck Index, 12th ed., Entry# 2795.
- Branco, L. C., et al., Studies on the selective transport of organic compounds by using ionic liquids as novel supported liquid membranes. Chemistry, 8(17), 3865-3871 (2002).
- Thomas, J. M., and Raja, R., Nanopore and nanoparticle catalysts. Chem. Rec., 1(6), 448-466 (2001).
- Raja, R., et al., Bifunctional molecular sieve catalysts for the benign ammoximation of cyclohexanone: one-step, solvent-free production of oxime and ε-caprolactam with a mixture of air and ammonia. J. Am. Chem. Soc., **123(33)**, 8153-8154 (2001).

GCY/NSB 5/06

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