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Product Information

β-Nicotinamide adenine dinucleotide 2'-phosphate reduced tetrasodium salt hydrate

≥97% (HPLC)

N7505

Product Description

CAS Registry Number: 2646-71-1

Synonyms: β -NADPH, Coenzyme II reduced tetrasodium salt, 2'-NADPH hydrate, NADPH, TPNH, Triphosphopyridine nucleotide reduced tetrasodium salt, NADPH Na₄, TPNH₂ Na₄, Dihydronicotinamide adenine dinucleotide phosphate tetrasodium salt

Molecular Formula: $C_{21}H_{26}N_7Na_4O_{17}P_3 \bullet xH_2O$

Formula Weight: 833.35 (anhydrous basis)

 E^{mM} (340nm) = 6.22 (pH >10)

 $A_{260nm} / A_{340nm} = 2.32$

Structure:



 β -Nicotinamide adenine dinucleotide 2'-phosphate (β -NADPH) is a product of the pentose phosphate pathway, a multifunctional pathway whose primary purpose is to generate reducing power, in the form of β -NADPH. β -NADPH transfers H⁺ and 2e⁻ to oxidized precursors in the reduction reactions of biosynthesis. Thus, β -NADPH cycles between catabolic and biosynthetic reactions, and serves as the carrier of reducing power in the same way that ATP serves as the carrier of energy.¹ Enzymes that use β -NADPH as a coenzyme include glutathione reductase, diacetyl reductase, dihydrofolate reductase, glutamic dehydrogenase, *p*-hydroxybenzoate hydroxylase, NADPH-FMN oxidoreductase, nitrate reductase and thioredoxin reductase. β -NADPH is also involved with cytochrome P450 electron transport systems.²

This listing of β -Nicotinamide adenine dinucleotide phosphate reduced form, N7505, is prepared by the chemical reduction of β -Nicotinamide adenine dinucleotide phosphate. Several theses³⁻⁵ and dissertations⁶⁻¹⁴ have cited use of N7505 in their research protocols.

Precautions and Disclaimer

For R&D use only. Not for drug, household, or other uses. Please consult the Safety Data Sheet for information regarding hazards and safe handling practices.

Storage/Stability

It is recommended to store N7505 desiccated at -20 °C, protected from light. The normal impurities and/or decomposition products are β -NADP and Monophosphoadenosine 5'-diphosphoribose.

It is suggested to prepare β -NADPH solutions fresh and use promptly, unless you are sure that this is an unnecessary precaution for your work. However, it has been reported that a 0.5 mM solution in 0.02 M NaOH (pH 12.3) showed no loss of purity in a week at 4 °C or -85 °C, but a 13% loss at -20 °C.¹⁵ One publication has investigated the solution stability of NADPH.¹⁶

Preparation Instructions

 $\beta\text{-NADPH}$ is tested for solubility in 0.01 M sodium hydroxide at 50 mg/mL.



References

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