Specification – Certified Reference Material

Certipur® Sodium hydrogen carbonate / sodium carbonate

Certified secondary standard reference material pH(S)=10.01₂ (25°C)

Accreditation:





Merck KGaA, Darmstadt, Germany is accredited by the German accreditation authority as registered reference material producer (D-RM-15185-01-00) in accordance with **ISO 17034**.

Producer: Merck KGaA, Frankfurter Str. 250, 64293 Darmstadt, Germany

Product no.: 1.01962.0001 (2.72071.0025 + 2.72072.0025) **Description of CRM:** Certipur® Sodium hydrogen carbonate / sodium carbonate

Certified secondary standard reference material for pH measurement directly traceable to

primary SRM from NIST/PTB pH(S)=10.01₂ (25°C)

Expiry date: 4 years

Storage: +15°C to +25°C tightly closed in the original container **Composition:** Sodium hydrogen carbonate / sodium carbonate

Formulation in compliance with DIN 19266, IUPAC, NIST, EP (Ph. Eur.) and USP

Temperature [°C]	Specification as pH (NaHCO $_3$ 0.025 mol/kg + Na $_2$ CO $_3$ 0.025 mol/kg)	Associated uncertainty U _{CRM} =k·u _{CRM} (k=2) as pH
5.0	10.240 - 10.260	± 0.010
10.0	10.168 - 10.188	± 0.010
15.0	10.106 - 10.126	± 0.010
20.0	10.051 - 10.071	± 0.010
25.0	10.002 - 10.022	± 0.010
30.0	9.959 – 9.979	± 0.010
37.0	9.903 - 9.923	± 0.010
40.0	9.880 - 9.900	± 0.011
45.0	9.844 - 9.864	± 0.011
50.0	9.813 - 9.833	± 0.011

Metrological traceability:

This certified secondary standard reference material is directly traceable to primary certified reference material sodium hydrogen carbonate / sodium carbonate characterised by PTB-CAR-xxx/xxxxx/xx and NIST xxx dx+xx.

PTB: Physikalisch Technische Bundesanstalt, Braunschweig, Germany NIST: National Institute of Standards and Technology, Gaithersburg, USA

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Method of analysis: The pH value is directly measured by differential potentiometry with the aid of two

platinum hydrogen electrodes "quasi without transference" according to IUPAC1 recommendations against solutions prepared from primary reference materials

characterised by PTB and NIST.

Intended use: This certified reference material is intended for use as a calibration standard for pH

instruments or pH electrodes or as a control sample for measuring the pH value.

Instructions for handling

The formulation is compliant to DIN 19266, IUPAC1, NIST2 and Ph. Eur. chapter 2.2.3 and USP chapter <791>.

and correct use: **Health and safety**

Please refer to the Safety Data Sheet for detailed information about the nature of

information: any hazard and appropriate precautions to be taken.

Details on correct use:

Preparation of sodium carbonate 0.025 mol/kg + sodium hydrogen carbonate 0.025 mol/kg $(pH(S)=10.01_2)$:

Dry sodium carbonate for 2 hours at 250°C and preserve over dry calcium chloride.

Dry sodium hydrogen carbonate for 2 days over molecular sieve.

Dissolve 2.640 g of sodium carbonate and 2.092 g of sodium hydrogen carbonate in 800 ml water and make up to 1000 ml at 25°C. The solution is sensitive to carbon dioxide.

Through within-bottle homogeneity a minimum sample volume of 30 ml was determined. The weigh-in quantity depends on the buffer substance and has to be calculated.

This certified reference material is intended for use as a calibration standard for pH instruments and pH electrodes. The pH value strongly depends on the temperature. Therefore it is necessary to keep the temperature constant during the measurement. Details concerning the nature of any hazard and appropriate precautions are provided in the material safety data sheet

Associated uncertainty:

The expanded uncertainty U_{CRM} is calculated as $U_{CRM} = k \cdot u_{CRM}$, where k=2 is the coverage factor for a 95% coverage probability and u_{CRM} is the combined standard uncertainty in accordance to ISO 17034.

The combined uncertainty u_{CRM} is derived from combination of the squared uncertainty contributions:

$$\mathbf{u}_{CRM} = \sqrt{\mathbf{u}^2}$$
Characterisation + \mathbf{u}^2 Homogeneity + \mathbf{u}^2 Stability

Ucharacterisation: is the uncertainty in accordance with DIN EN ISO/IEC 17025 which includes the

contributions of the primary reference material and the measuring system. The characterisation measurements have been conducted by our DAkkS accredited

calibration laboratory.

is the between-bottle variation in accordance with ISO 17034. The assessment Uhomogeneity:

of homogeneity is performed by analysis of a representative number of

systematically chosen sample units.

is the uncertainty obtained from short-term and long-term stability in accordance Ustability:

with ISO 17034. The stability studies are the basis for the quantification of the

expiry date of this reference material for the unopened bottle.

Detailed information is provided by the certificates and the certification report on our website.

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¹ R.P. Buck, et al.: The Measurement of pH - **Definition, Standards and Procedures (IUPAC Recommendations 2002)**, Pure Appl. Chem, Vol 74, No. 11, pp. 2169-2200, 2002

² Y. Ch. Wu, W. F. Koch, R. A. Durst: **Standardization of pH Measurements**, NBS Special Publication 260-53, 1988