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# **Product Information**

# Thrombin from human plasma

Catalog Number **T6884** Storage Temperature –20 °C

CAS RN 9002-04-4 EC 3.4.21.5

Synonyms: Factor IIa, FIIa, fibrinogenase, thrombase, tropostasin, activated blood-coagulation factor II EXPASY/SwissProt P00734

# **Product Description**

Thrombin is an endolytic serine protease that selectively cleaves the Arg–Gly bonds of fibrinogen to form fibrin and release fibrinopeptides A and B.  $^{1,2}$  The predominant form of thrombin *in vivo* is the zymogen prothrombin (factor II), which is produced in the liver. The concentration of prothrombin in normal human plasma is 5–10 mg/dL. Prothrombin is a glycoprotein with a glycan content of ~12%. Prothrombin is cleaved *in vivo* by activated factor X (factor Xa), releasing the activation peptide and cleaving thrombin into light and heavy chains, which yields catalytically active  $\alpha$ -thrombin.

 $\alpha$ -Thrombin is composed of a light chain (A chain, MW  $\sim$ 6 kDa) and a heavy chain (B chain, MW  $\sim$ 31 kDa). These two chains are joined by one disulfide bond. The B chain of human thrombin consists of a peptide portion (MW 29,485 Da) and a carbohydrate portion (MW 2,334 Da) with N-linked glycosylation at three Asn residues. Human thrombin has been reported to contain 4.1% hexose, 1.7% sialic acid, and  $\sim$ 2.4% acetylglucosamine.  $^{5,6}$ 

Autolytic degradation of  $\alpha$ -thrombin results in the formation of  $\beta$ - and  $\gamma$ -thrombin. These catalyze cleavage of chromogenic, synthetic substrates, but have lower fibrinogen clotting activity.  $\beta$ -Thrombin is formed from  $\alpha$ -thrombin by degradation of the A chain and the excision of a small fragment containing a carbohydrate from the B chain.<sup>3</sup>

Thrombin also contains  $\gamma$ -carboxyglutamyl residues. These modified glutamyl residues are the result of carboxylation by vitamin K-dependent carboxylase, a microsomal enzyme.  $\gamma$ -Carboxyglutamyl residues are necessary for the Ca<sup>2+</sup>-dependent interaction with a negatively charged phospholipid surface, which is essential for the conversion of prothrombin to thrombin.

Prothrombin is activated *in vivo* on the surface of a phospholipid membrane that binds the N-terminus of prothrombin along with factors Va and Xa. The activation process starts slowly because factor V is activated to factor Va by the initial, small amount of thrombin.

The optimal cleavage sites for thrombin are as follows:1

- A-B-Pro-Arg-||-X-Y, where A and B are hydrophobic amino acids, and X and Y are nonacidic amino acids
- 2. Gly-Arg-||-Gly

Thrombin cleavage of fibrinogen occurs only at Arg residues. However, the cleavage is not site-specific, and generally results in 2 products:

- The primary cleavage product, fibrinopeptide A, is cleaved from fibrinogen after amino acid 16 and sometimes after amino acid 19.
- A secondary cleavage product, fibrinopeptide B, is produced by cleavage at amino acid 14.<sup>7</sup>

Thrombin from any mammalian species will clot the fibrinogen of any other mammalian species. Thrombin does not require divalent metal ions or cofactors for activity. However, Na<sup>+</sup>-dependent allosteric activation of thrombin has been shown to play a role in defining the primary specificity of thrombin to cleave after Arg residues. \*\*

Thrombomodulin serves as a cofactor for thrombin during the activation of protein C. Thrombin (human and bovine) will catalyze the hydrolysis of several peptide p-nitroanilides, tosyl-Arg-nitrobenzyl ester, and thiobenzyl ester synthetic substrates.  $^{10}$ 

Catalytic pH range:  $^{11}$  5–10 Optimal pH:  $^{11}$  8.3 (Note: thrombin precipitates at pH  $\leq$ 5) Molecular mass:  $^{4,12}$  37.4 kDa Human isozymes pl range: 6.35–7.6  $E_{280}^{1\%}$  = 18.3 (human)  $^{12}$  Thrombin can also be used to cleave fusion proteins. Cleavage of fusion proteins can be carried out at a thrombin:fusion protein ratio of 1:500.  $^{13}$  A concentration of 0.5 NIH units thrombin per one nanomole of polypeptide in 20  $\mu$ L of 50 mM ammonium bicarbonate, pH 8.0, has also been described.  $^{2}$ 

This product is lyophilized from a solution of saline sodium citrate buffer, pH 6.5.

Specific Activity:  $\geq 2,000$  NIH units/mg protein ( $E_{280}^{1\%} = 18.3$ )

Unit Definition: Activity is expressed in NIH units obtained by direct comparison to an NIH Thrombin Reference Standard. The NIH assay procedure uses 0.2 mL of diluted plasma (1:1 with saline) as a substrate and 0.1 mL of albumin solution based on a modification of the method of Biggs. Only clotting times in the range of 15–25 seconds are used for determining thrombin activity. Optimal clotting temperature is 37 °C.

Thrombin concentrations in the literature are typically reported in terms of different units of activity. <sup>14,15</sup> Several conventions are used to express thrombin activity in the literature:

1 IOWA unit = 0.83 NIH unit

1 WHO unit = 1 NIH unit

1 NIH unit =  $0.324 \pm 0.073 \ \mu g$ 

1 NIH unit = 1 USP unit

# **Precautions and Disclaimer**

This product is for R&D use only, not for drug, household, or other uses. Please consult the Safety Data Sheet for information regarding hazards and safe handling practices.

## **Preparation Instructions**

The product is soluble in water (10 mg/mL).

## Storage/Stability

Store the lyophilized powder at –20 °C. The product retains activity for at least 5 years.

Stock solutions can be prepared at a concentration of 100 units/mL in a 0.1% (w/v) BSA solution. Stock solutions remain active for one week at 0–5 °C. Solutions are most stable at pH 6.5, as a pH >7 will greatly reduce thrombin activity. Since thrombin solutions adsorb to glass, it is recommended to aliquot the solutions in plastic tubes and store at -20 °C for long-term storage.

### References

- Chang, J.Y., Eur. J. Biochem., 151(2), 217-224 (1985).
- Doolittle, R.F., in *The Plasma Proteins*, Volume II (Biosynthesis Metabolism, Alterations in Disease), 2nd ed. (Putnam, F. W., ed.). Academic Press (New York, NY), pp. 148-149 (1975).
- 3. Qian, W.J., et al., J. Proteome Res., 4, 2070-2080 (2005).
- 4. Nilsson, B., et al., Arch. Biochem. Biophys., **224(1)**, 127-133 (1983).
- Magnusson, S., in *The Enzymes* (Third Edition), Vol. III (Boyer, P.D., ed.). Academic Press (New York, NY), pp. 277-321 (1971).
- 6. Lanchatin, G.F., et al., J. Biol. Chem., **243(20)**, 5479-5488 (1968).
- 7. Machovich, R. (ed.), *The Thrombin*, Vol. 1. CRC Press (Boca Raton, FL), pp. 63-66 (1984).
- 8. Prasad, S., *J. Biol. Chem.*, **279**, 10103-10108 (2004).
- 9. Kisiel, W., J. Clin. Invest., **64(3)**, 761-769 (1979).
- 10. Lottenberg, R., et al., Meth. Enzymol., **80(Part C)**, 341-361 (1981).
- 11. Machovich, R. (ed.), *The Thrombin*, Vol. 1, CRC Press (Boca Raton, FL), p. 111 (1984).
- 12. Butkowski, R.J., et al., J. Biol. Chem., **252(14)**, 4942-4957 (1977).
- 13. Hakes, D.J., and Dixon, J.E., *Anal. Biochem.*, **202(2)**, 293-298 (1992).
- Biggs, R., ed., Human Blood Coagulation, Haemostasis and Thrombosis (2<sup>nd</sup> ed.), Blackwell Scientific Publications (Philadelphia, PA), p. 722 (1976).
- 15. Hemker, H.C., Handbook of Synthetic Substrates for the Coagulation and Fibrinolytic System, Martinus Nijhoff (Boston, MA) / Springer (Dordrecht, The Netherlands), pp. 95-101 (1983).

RBG,TMG,RXR,GCY,MAM 12/17-1