

Lyophilized powder, 3-8 TIU/mg solid

A1153

CAS Registry Number: 9087-70-1

Synonyms: Basic pancreatic trypsin inhibitor, Bovine pancreatic trypsin inhibitor (BPTI), Kallikrein-trypsin inactivator, Kunitz protease inhibitor

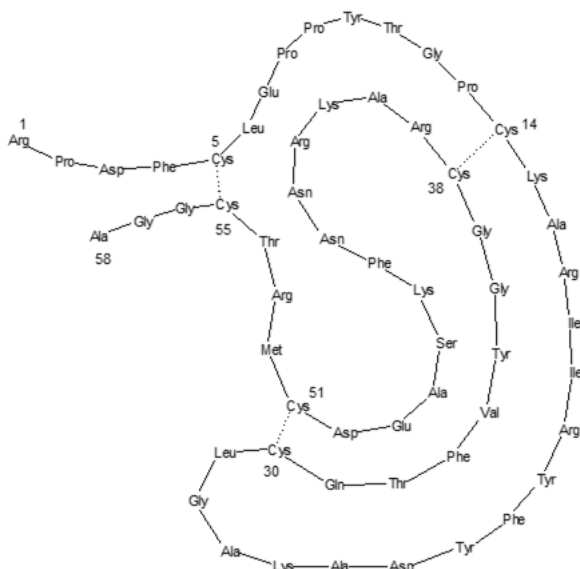
Molecular formula: C₂₈₄H₄₃₂N₈₄O₇₉S₇

Molecular weight: 6,512 Da

pI: 10.5

Extinction coefficient: $E^{1\%} = 8.4$ (280 nm, water)

Aprotinin is a protein consisting of 58 amino acids, arranged in a single polypeptide chain that is cross-linked by three disulfide bridges (depicted below). Aprotinin is a competitive serine protease inhibitor that forms stable complexes with and blocks the active sites of various enzymes, such as those listed in Table 1. The binding is reversible. Most aprotinin-protease complexes dissociate at pH >10 or pH < 3.



Several theses¹⁻³ and dissertations⁴⁻¹⁶ have cited use of product A1153 in their protocols.

Inhibitory Activity: 3-8 TIU/mg solid

Unit Definition: One Trypsin Inhibitor Unit (TIU) will decrease the activity of 2 trypsin units by 50%, where 1 trypsin unit will hydrolyze 1.0 μ mole of *N*- α -benzoyl-DL-arginine *p*-nitroanilide (BAPNA) per minute at pH 7.8 and 25 $^{\circ}$ C.

Another commonly used unit of activity is the KIU (Kallikrein Inhibitor Unit). Different conversion factors between aprotinin units have been reported:

- 1 TIU = 1,300 KIU
- 1 TIU = 1,025 KIU¹⁷

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Preparation Instructions

Aprotinin is freely soluble in water (5 mg/mL) and in aqueous buffers of low ionic strengths.¹⁸

Storage/Stability

Store the lyophilized powder at 2-8 °C. When stored at 2-8 °C, the product retains activity for at least 2 years.

Dilute solutions of aprotinin are generally less stable than concentrated ones. Solution stability is pH-dependent, although a pH range of 1-12 can be tolerated.¹⁹ Repeated freeze-thaw cycles should be avoided.

The Cys¹⁴-Cys³⁸ disulfide bridge is readily split by reducing agents like 2-mercaptoethanol.¹⁹ Due to its compact tertiary structure, aprotinin is relatively stable against denaturation due to high temperature, organic solvents, or proteolytic degradation (See Table 2). Only thermolysin has been found capable of degrading aprotinin after heating to 60-80 °C.¹⁹

The high basicity of aprotinin causes it to adhere to commonly used dialysis tubing and even gel filtration matrices. However, the use of acetylated materials and concentrated salt solutions (such as ≥ 0.1 M NaCl in buffer) minimizes this problem.¹⁹

Sterilization may be achieved by filtration through a 0.2 µm filter.

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Table 1. Inhibition by Aprotinin

Enzyme (Source), Condition	Inhibition (K_i)
Acrosin	Weak inhibition ¹⁸
Chymotrypsin	$K_i = 9 \text{ nM}^{20}$
CMP- <i>N</i> -Acetyl-neuraminate lactosylceramide α -2,3-sialyltransferase	74% Inhibition at 300 nM ²⁰
Elastase (human leukocyte), pH 8.0	$K_i = 3.5 \text{ } \mu\text{M}^{19}$
Kallikrein (pancreatic), pH 8.0	$K_i = 1.0 \text{ nM}^{19}$
Kallikrein (plasma)	$K_i = 30 \text{ nM}$; 100 nM ¹⁸
Kallikrein (tissue)	$K_i = 1 \text{ nM}^{18}$
Kallikrein (urine)	$K_i = 1.7 \text{ nM}^{18}$
Plasmin (porcine), pH 7.8	$K_i = 4.0 \text{ nM}^{19}$
Plasminogen activator	$K_i = 8 \text{ } \mu\text{M}$; 27 μM^{18}
Trypsin (bovine), pH 8.0	$K_i = 0.06 \text{ pM}^{19}$
Trypsinogen (bovine), pH 8.0	$K_i = 1.8 \text{ } \mu\text{M}^{19}$
Tryptase TL-2	16% Inhibition at 10 μM^{18}
Urokinase (human), pH 8.8	$K_i = 8.0 \text{ } \mu\text{M}^{19}$

Table 2. Aprotinin solution stability

Solvent	Concentration	Temperature	% Loss/Time
Sterile water with 0.9% NaCl and 0.9% benzyl alcohol, pH 5.7-6.2	10 mg/mL	0-5 °C	< 4.3%/year
2.5% Trichloroacetic acid	N/A	80 °C	No loss ²¹
pH < 12.6	N/A	N/A	No loss observed after 24 hours ²²
pH < 12	N/A	N/A	Irreversibly denatured ²³
pH 7-8	0.065-1.95 $\mu\text{g/mL}$	4 °C	About 1 week ¹⁸
pH 7-8	0.065-1.95 $\mu\text{g/mL}$	-20 °C	> 6 months ¹⁸

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