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ProductInformation

N-tert-Butyl-α-phenylnitrone

Product Number **B 7263** Storage Temperature -0 °C

Product Description

Molecular Formula: C₁₁H₁₅NO Molecular Weight: 177.2 CAS Number: 3376-24-7 Melting Point: 74-75 C

N-tert-butyl-a-phenylnitrone (PBN) is a commonly used free-radical spin trapping agent. 1-3 The technique of spin trapping is used in electron spin resonance (ESR) and makes use of a diamagnetic compound (the spin trap), such as PBN, which reacts with a free radical (the spin), giving rise to a relatively stable ESRobservable free radical (spin adduct).³ PBN has been shown to reduce the number of emboli-induced cerebral microinfarctions in the rabbit cortex and prevent neoplasia by its radical scavenging activity and its abilty to inhibit cyclooxygenase-2 activity at the catalytic level.^{4,5} PBN in a dose of 100 mg/kg i.p. reduced necrosis of the substantia nigra, pars reticulate in flurothyl-induced status epilepticus in rats. The lethal dose of PBN in rats was found to be approximately 100 mg/100 g body weight (0.564 mmol/100g).^{6,7} PBN has also been used to generate nitric oxide via light-induced decomposition⁸ and to investigate free radical generation during cardiac dysfunction. There was a linear, direct relationship between the magnitude of the PBN adduct production and the degree of ischemic flow reduction.⁹

Precautions and Disclaimer

For Laboratory Use Only. Not for drug, household or other uses.

Preparation Instructions

PBN is soluble in DMSO. For infusion into animals, a 5 mg/ml solution was prepared in a solution of 80% normal saline and 20% water. The product is also soluble in chloroform (50 mg/ml).

Storage/Stability

PBN is sensitive to light, especially in solution. 10

References

- Kalyanaraman, B., et al., Site-specific trapping of reactive species in low-density lipoprotein oxidation: biological implications. Biochim. Biophys. Acta., 1168, 220-227 (1993).
- Mullins, M.E., EPR Spectroscopic Characterization of Biological Thiyl Radicals as PBN Spin-Trap Adducts. Applied Magnetic Resonance, 3, 1021-1032 (1992).
- 3. Evans, C.A., Spin Trapping. Aldrichimica Acta, **12**, 23-30 (1979).
- Roos, M.W., and Ericsson, A., N-tert-butyl-alphaphenylnitrone reduces the number of microinfarction in the rabbit brain cortex. Exp. Brain. Res. 124, 251 (1999).
- Nakae, D., Inhibition by phenyl-N-tert-butylnitrone of early phase carcinogenesis in the livers of rats fed a choline-deficient, L-amino acid-defined diet. Cancer Res., 58, 4548 (1998).
- He, Q.P., Necrosis of the substantia nigra, pars reticulate, in flurothyl-induced status epilepticus is ameliorated by the spin trap α-phenyl-N-tert-butyl nitrone. Free Radical Biol. Med., 22, 917 (1997).
- 7. Janzen, E.G., Biological spin trapping. II. Toxicity of nitrone spin traps:dose-ranging in the rat. J. Biochem. Biophys. Meth. **30**, 239 (1995).
- 8. Chamulitrat, W., et al., Nitric Oxide Formation during Light-induced Decomposition of Phenyl N-tert-Butylnitrone. J. Biol. Chem., **268**, 11520-11527 (1993).

- 9. Bolli, R., et al., Demonstration of Free Radical Generation in "Stunned" Myocardium of Intact Dogs with the Use of the Spin Trap α -Phenyl
- N-Tert-Butyl Nitrone. J. Clin. Invest., **82**, 476-485 (1988).
- 10. The Merck Index, 13th ed., Entry # 7129.

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